

## Synthesis of hedychenone, yunnancoronarins and aframodial derivatives

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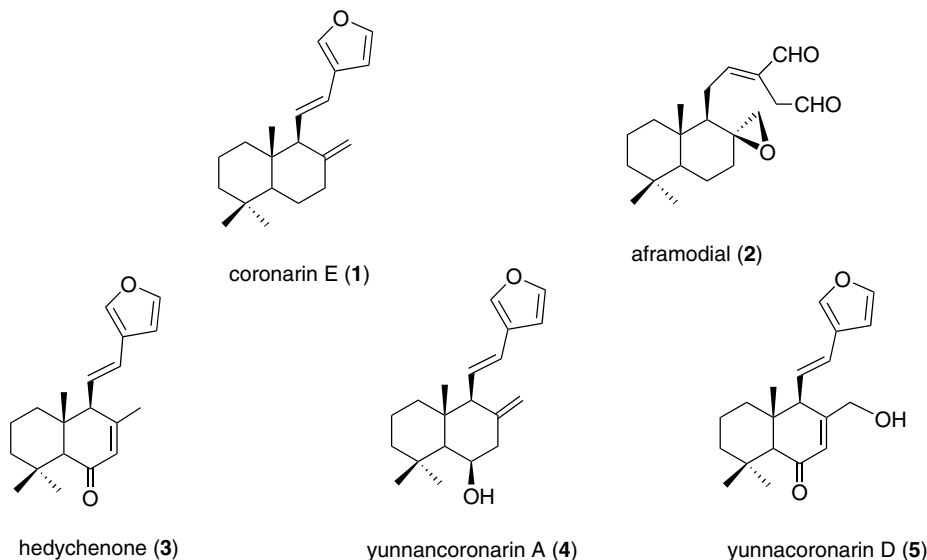
This paper is dedicated to the memory of F. Aslaoui

**Abstract**—Starting with larixol, syntheses of furan and 1,4-enedial labdane-type diterpenes are presented, which has enabled a preparation of hedychenone (20% yield).  
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Extracts of Zingiberaceae, one of the major tropical plant families, have long been used in traditional medicine<sup>1</sup> and their examination has resulted in the isolation of labdanes such as the biogenetically-related coronarin E (**1**) and aframodial (**2**).<sup>2,3</sup> In particular, the rhizomes<sup>4</sup> of *Hedychium spicatum*, *H. yunnanense*, *H. coronarium* and *H. forrestii* were found to contain bioactive hedychenone (**3**) yunnancoronarins A (**4**) and D (**5**);<sup>5–7</sup> the structures of these labdanes have been established by spectroscopic

methods<sup>8</sup> and that of hedychenone confirmed by radio-crystallographic analysis;<sup>9</sup> in this work we disclose the synthesis of **3–5** as well as a compound related to **2**.

At first larixol (**6**), a labdane which is readily available from larch oleoresin,<sup>10</sup> was converted into the aldehyde **7**<sup>11</sup> by a three-steps transformation based on an osmium tetroxide/sodium periodate oxidation sequence;<sup>12</sup> protection of the hydroxyl group as a silyl ether was



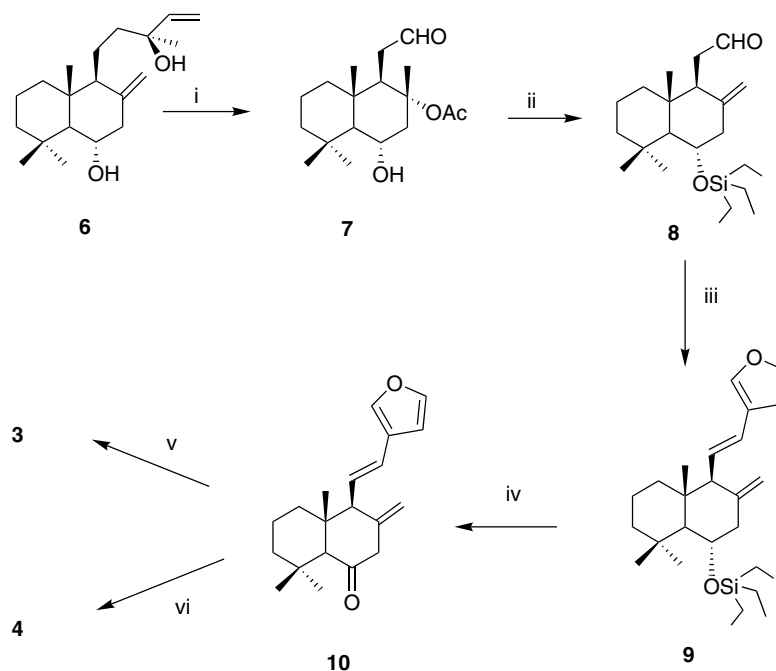
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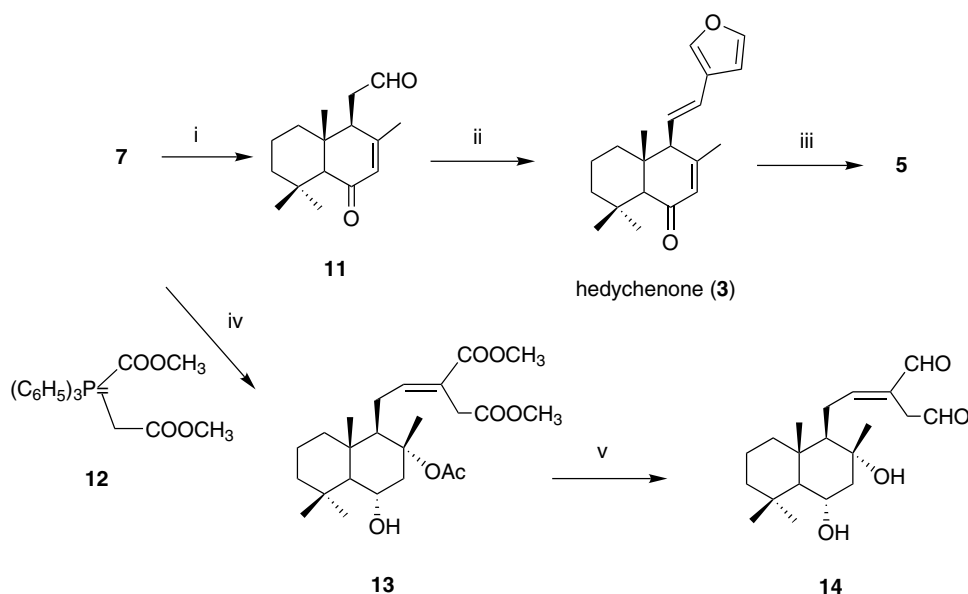
followed by elimination of the tertiary acetate by use of 2,4,6-collidine;<sup>13</sup> that the double bond was *exo* was revealed by the presence of two vinylic protons (singlets at 4.8 and 4.55 ppm). Then, addition of 3-furyl lithium<sup>14</sup> to the aldehydic group of **8** afforded a ca. 1:1 mixture of epimeric alcohols, which without separation, were mesylated in the presence of 2,6-lutidine; this gave the elimination product **9** with the desired *trans*-configuration ( $J = 16$  Hz). Cleavage of the silyl ether followed by oxi-

dation then afforded **10**. This intermediate gives access to both hedychenone (**3**), after isomerization, and yunnanconarin A (**4**) after reduction with diisobutylaluminum hydride which, for steric reasons, occurs from the alpha side (Scheme 1).

With a preparation of hedychenone in hand, a shorter route based on the preferential reaction of an aldehyde group (over a ketone) was explored (Scheme 2); after



**Scheme 1.** Reagents and conditions: (i) Ref. 11 (50%); (ii) (1) Et<sub>3</sub>SiCl, DMAP cat, py, 8 h (94%), (2) 2,4,6-collidine as solvent, 170 °C, 12 h (79%); (iii) (1) 3-furyllithium, −78 °C, 2 h (68%), (2) 2,6-lutidine (7 equiv), CH<sub>2</sub>Cl<sub>2</sub>, add MsCl (3 equiv), rt, 18 h (68%); (iv) (1) AcOH, THF–H<sub>2</sub>O (5:1:3), rt, overnight (95%), (2) IBX (3 equiv) AcOEt, 60 °C, 3 h (85%); (v) MeONa (0.2 M in MeOH), 2 h (quant); (vi) THF, −78 °C, Dibal-H (6 equiv), 2 h (95%).



**Scheme 2.** Reagents and conditions: (i) (1) IBX (3 equiv), AcOEt, 70 °C, 3 h (84%); (2) 2,4,6-collidine, 160 °C, 2 h (90%); (ii) (1) 3-furyllithium, −78 °C, (2) 2,6-lutidine, MsCl (59%, two steps); (iii) (1) SeO<sub>2</sub> (1.5 equiv), dioxane, 80 °C—8 h; (2) NaBH<sub>4</sub>, EtOH, −78 °C (60% two steps); (iv) **12** (65%); (v) LAH, then IBX (56% overall), see Ref. 21.

oxidation of **7** followed by conjugation of the *exo* double bond, enone **11** was obtained.<sup>15</sup> Pleasingly, the addition of 3-furyl lithium occurred preferentially on its aldehyde group to afford **3** after mesylation/elimination.<sup>16</sup> Hedychenone (**3**), identical to that prepared according to Scheme 1, is now obtained in four steps from **7** (22% overall yield from larixol).

Yunnancoronarin D (**5**) could be synthesized from hedychenone in only two steps: oxidation of the allylic methyl group, followed by reduction of the aldehyde group thus formed (Scheme 2); as other yunnancoronarins have been obtained by photooxygenation<sup>17</sup> of hedychenone, this work also constitutes their formal preparation. Finally, the availability of **7** led us to consider introduction of the 1,4-enedial (3-formyl-3-butenal) structural unit as this pharmacophore is found in a number of bioactive terpenoids.<sup>18</sup> Reaction of ylide **12**<sup>19</sup> with **7** afforded **13** (only the *E* isomer was obtained);<sup>20</sup> reduction of **13**, followed by selective oxidation<sup>21</sup> then afforded the desired 1,4-enedial **14**<sup>22</sup> (Scheme 2).

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#### References and notes

- Pancharoen, O.; Prawat, U.; Tuntiwachwuttikul, P. *Stud. Nat. Prod. Chem.* **2000**, 23(Part D), 797–865.
- Jung, M.; Ko, I.; Lee, S. *J. Nat. Prod.* **1998**, 61, 1394–1396; Jung, M.; Ko, I.; Lee, S.; Choi, S. J.; Youn, B. H.; Kim, S. K. *Bioorg. Med. Chem. Lett.* **1998**, 8, 3295–3298; Müller, M.; Schröder, J.; Magg, C.; Seiffert, K. *Tetrahedron Lett.* **1998**, 39, 4655–4656; Nozawa, M.; Ono, E.; Akita, H. *Heterocycles* **2000**, 53, 1811–1819; Villamizar, J.; Fuentes, J.; Salazar, F.; Tropper, E.; Alonso, R. *J. Nat. Prod.* **2003**, 66, 1623–1627; Villamizar, J.; Salazar, F.; Fuentes, J.; Tropper, E.; Alonso, R. *J. Chem. Res. (S)* **2003**, 504–506; Oh, S.; Jeong, I. H.; Shin, W.-S.; Lee, S. *Bioorg. Med. Chem. Lett.* **2003**, 13, 2009–2012.
- Itokawa, H.; Morita, H.; Katou, I.; Takeya, K.; Cavalheiro, A. J.; De Oliveira, R. C. B.; Ishige, M.; Motimode, M. *Planta Med.* **1988**, 54, 311–315; Nyasse, B.; Lenta-Ndjakou, B. *Pharmazie* **2000**, 55, 703–704.
- Srimal, R. C.; Sharma, S. C.; Tandon, J. S. *Ind. J. Pharmacol.* **1984**, 16, 143–147; For a review about chemical and pharmacological studies on diterpenoids from Hedychium see: Zhao, Q.; Zou, C.; Hao, X.; Chen, Y. *Tianran Chanwu Yanjiu Yu Kaifa* **1999**, 11, 85–91.
- Morikawa, T.; Matsuda, H.; Sakamoto, Y.; Ueda, K.; Yoshikawa, M. *Chem. Pharm. Bull.* **2002**, 50, 1045–1049; Matsuda, H.; Morikawa, T.; Sakamoto, Y.; Toguchida, I.; Yoshikawa, M. *Heterocycles* **2002**, 56, 45–50.
- Zhao, Q.; Hao, X. J.; Chen, Y. Z.; Zou, C. *Acta Pharm. Sinica* **1995**, 30, 119–122.
- Matsuda, H.; Morikawa, T.; Sakamoto, Y.; Toguchida, I.; Yoshikawa, M. *Bioorg. Med. Chem.* **2002**, 10, 2527–2534.
- Hedychenone: Sharma, S. C.; Tandon, J. S.; Uprety, H.; Shukla, Y. N.; Dhar, M. M. *Phytochemistry* **1975**, 14, 1059–1061; Yunnancoronarin A: Zhao, Q.; Hao, X.-J.; Chen, Y.-Z.; Zou, C. *Chem. J. Chin. Univ.* **1995**, 16, 64–68; Zhao, Q.; Hao, X.-J.; Chen, Y.-Z.; Zou, C.; Hong, X. *Acta Bot. Sinica* **1999**, 41, 528–530; Yunnancoronarin D: Zou, C.; Zhao, Q.; Hao, X.; Chen, Y.; Hong, X. *Acta Bot. Yunnanica* **1999**, 21, 253–255.
- Liu, X.-H.; Wang, W.; He, Z.-D.; Wang, H.-Q.; Yang, C.-R.; Chen, B. *Acta Crystallogr. Sect. C, Cryst. Struct. Commun.* **1999**, C55.
- Mills, J. S. *Phytochemistry* **1973**, 12, 2407–2412.
- Lagnol, B. M. F.; Morin, C.; de Groot, A. *Synthesis* **2000**, 1907–1916; Bolster, M. G.; Jansen, B. J. M.; de Groot, A. *Tetrahedron* **2001**, 57, 5663–5679.
- The work of S. Proskow on the oxidative degradation of sclareol has remained unpublished but is quoted in: Corey, E. J.; Sauers, R. R. *J. Am. Chem. Soc.* **1959**, 81, 1739–1743; for a mechanism see: Barrero, A. F.; Alvarez-Manzaneda, E. J.; Altarejos, J.; Salido, S.; Ramos, J. M. *Tetrahedron* **1993**, 49, 10405–10412.
- Barrero, A. F.; Manzaneda, E. A.; Altarejos, J.; Salido, S.; Ramos, J. M.; Simmonds, M. S. J.; Blaney, W. M. *Tetrahedron* **1995**, 51, 7435–7450.
- Prepared by lithiation (*n*-BuLi, THF, –78 °C, 15 min) of 3-bromofuran, which is expensive but can be prepared; see: Kraus, G. A.; Wang, X. *Synth. Commun.* **1998**, 28, 1093–1096.
- This material should be prepared freshly before use; characteristic spectroscopic data: <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ: 9.89 (s, 1H, CHO), 5.78 (large s, 1H, H-7); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 200.8 (C=O); 199.4 (CHO); 155.9 (C-8); 128.9 (C-7).
- This step was carried out as in Scheme 1 (e.g., **8** to **9**) but without purification of the intermediates.
- Zhao, Q.; Hao, X. J.; Chen, Y. Z.; Zou, C. *Chin. Chem. Lett.* **1996**, 7, 25–28.
- For a review on biologically active unsaturated 1,4-dialdehydic terpenoids, see: Jonassohn, M.; Sterner, O. *Trends Org. Chem.* **1997**, 6, 23–43.
- Compennolle, F.; Joly, G.; Peeters, K.; Toppet, S.; Hoornaert, G.; Kilonda, A.; Babady-Bila *Tetrahedron* **1997**, 53, 12739–12754; For use of this ylide with a drimane keto aldehyde, see: Kim, T. H.; Isoe, S. *J. Chem. Soc., Chem. Commun.* **1983**, 730–731.
- This configuration was assigned after observation of a NOE effect between H-11 and H-14.
- IBX/AcOEt was used under the conditions described by: More, J. D.; Finney, N. S. *Org. Lett.* **2002**, 4, 3001–3003, 6-keto by-products are obtained but the overall yield of the reaction scheme is fair enough to avoid a protection/deprotection sequence of the secondary hydroxyl group (OH-6).
- Under Ar, to a solution of **7** (167 mg, 0.539 mmol) in dry toluene (3 mL) was added ylide **12**<sup>18</sup> (875 mg, 2.16 mmol, 4 equiv) and the mixture was heated at 115 °C for 72 h. After cooling and evaporation of the volatiles the residue was purified by column chromatography on silica gel (cyclohexane/AcOEt 8:2) to afford **13** (oil, 158 mg, 0.361 mmol, 67%) which was then added as a THF solution (2 mL) to a suspension of LiAlH<sub>4</sub> (10 equiv) in dry THF (4 mL) at 4 °C. After 12 h stirring at rt the solution was cooled to 4 °C and 1 M aqueous HCl was added before extractive work-up with ethyl acetate. After evaporation of the volatiles, the residue was taken-up in ethyl acetate (4 mL) and IBX (prepared according to Frigerio, M.; Santagostino, M. *Tetrahedron Lett.* **1994**, 35, 8019–8022) (4 equiv) added before stirring 4 h at 60 °C. After cooling and evaporation of the volatiles, column chromatography (cyclohexane/AcOEt 6:4) afforded **14** in 56% yield. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ: 9.69 (s, 1H, H-16); 9.45 (large s, 1H, H-15); 7.01 (t, *J*<sub>14–15</sub> = 7 Hz, 1H,

H-12); 4.10 (m, 1H, H-6); 3.45 and 3.55 (AB system,  $J_{AB} = 16$  Hz, 2H, H-14); 1.53 (s, CH<sub>3</sub>-17); 1.21, 0.95, 0.84 (3 s, CH<sub>3</sub>-18, -19, -20). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 198.3 (C-16); 193.3 (C-15); 160.4 (C-12); 133.8 (C-13); 76.0 (C-8); 66.4 (C-6); 62.2 (C-9, C-5); 42.5 (C-7); 40.8 (C-3);

39.4 (C-1); 39.0 (C-10); 35.8 (C-14); 32.3 (C-18); 29.6 (C-4); 25.9 (C-11); 22.4, 21.9 (C-19, C-17); 18.0 (C-2); 16.3 (C-20). For data on aframodial: see: Kimbu, S. F.; Njimi, T. K.; Sondengam, L.; Akinniyi, J. A.; Connolly, J. D. *J. Chem. Soc., Perkin Trans. I* **1979**, 1303–1305.